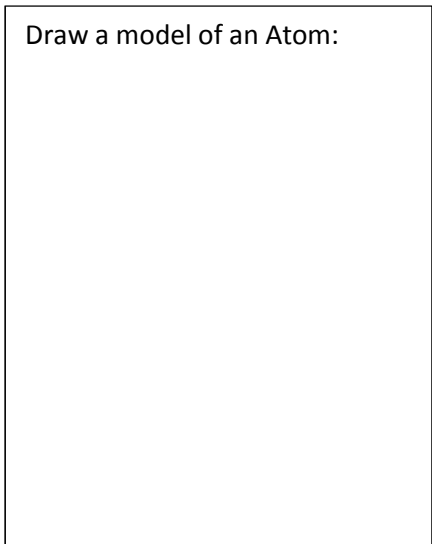


Science 8 - The Atom – notes

Name: _____

Particle	Location	Charge	Mass
Proton			
Neutron			
Electron			

Draw a model of an Atom:



Inquiry Activity:

- A. Take a package of 20 elements.
- B. As a group decide how to put them in order.
- C. Once they are in order - find a **property** that allows you to divide them into 4 groups.
- D. Show me once you feel you have solved the puzzle

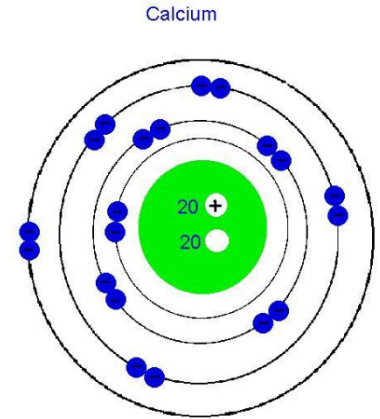
Questions to answer about 20 elements:

- 1. What happens to the number of protons as you move from smallest to biggest?
- 2. What happens to the number of neutrons as you move from smallest to biggest?
- 3. Are the number of protons and neutrons always the same?
- 4. What happens to the number of electrons as you move from smallest to biggest?
- 5. Which particles are found in the nucleus?
- 6. Which particle in the nucleus always has the same amount as the number of electrons?

Orbitals:

- Describe how are the electrons organized?
- What is the maximum number of electrons found in each orbit?

Orbit	Max # of Electrons
1	
2	
3	
4	



- Relate this organization to the four groups that you made?

KEY

- Alkali metals
- Alkali-earth metals
- Transition metals
- Rare earths
- Radioactive rare earths
- Other metals
- Semimetals
- Non-metals
- Noble gases
- Hydrogen

1 H Hydrogen 1																	2 He Helium 4						
3 Li Lithium 7	4 Be Beryllium 9																	5 B Boron 11	6 C Carbon 12	7 N Nitrogen 14	8 O Oxygen 16	9 F Fluorine 19	10 Ne Neon 20
11 Na Sodium 23	12 Mg Magnesium 24																	13 Al Aluminum 27	14 Si Silicon 28	15 P Phosphorus 31	16 S Sulphur 32	17 Cl Chlorine 35	18 Ar Argon 40
19 K Potassium 39	20 Ca Calcium 40	21 Sc Scandium 45	22 Ti Titanium 48	23 V Vanadium 51	24 Cr Chromium 52	25 Mn Manganese 55	26 Fe Iron 56	27 Co Cobalt 59	28 Ni Nickel 58	29 Cu Copper 63	30 Zn Zinc 64	31 Ga Gallium 69	32 Ge Germanium 74	33 As Arsenic 75	34 Se Selenium 80	35 Br Bromine 79	36 Kr Krypton 84						
37 Rb Rubidium 85	38 Sr Strontium 88	39 Y Yttrium 89	40 Zr Zirconium 90	41 Nb Niobium 93	42 Mo Molybdenum 98	43 Tc Technetium 97	44 Ru Ruthenium 102	45 Rh Rhodium 103	46 Pd Palladium 106	47 Ag Silver 107	48 Cd Cadmium 114	49 In Indium 115	50 Sn Tin 120	51 Sb Antimony 121	52 Te Tellurium 130	53 I Iodine 127	54 Xe Xenon 132						
55 Cs Caesium 133	56 Ba Barium 138	57-71	72 Hf Hafnium 180	73 Ta Tantalum 181	74 W Tungsten 184	75 Re Rhenium 187	76 Os Osmium 192	77 Ir Iridium 193	78 Pt Platinum 195	79 Au Gold 197	80 Hg Mercury 202	81 Tl Thallium 205	82 Pb Lead 208	83 Bi Bismuth 209	84 Po Polonium 209	85 At Astatine 210	86 Rn Radon 222						
87 Fr Francium 223	88 Ra Radium 226	89-103	104 Unq Unnilquadium 260	105 Unp Unnilpentium 262	106 Unh Unnilhexium 263	107 Uns Unnilseptium 262	108 Uno Unniloctium 265	109 Uue Unnilennium 266															
57 La Lanthanum 139	58 Ce Cerium 140	59 Pr Praseodymium 141	60 Nd Neodymium 142	61 Pm Promethium 145	62 Sm Samarium 152	63 Eu Europium 153	64 Gd Gadolinium 158	65 Tb Terbium 159	66 Dy Dysprosium 164	67 Ho Holmium 165	68 Er Erbium 168	69 Tm Thulium 169	70 Yb Ytterbium 174	71 Lu Lutetium 175									
89 Ac Actinium 227	90 Th Thorium 232	91 Pa Protactinium 231	92 U Uranium 238	93 Np Neptunium 237	94 Pu Plutonium 244	95 Am Americium 243	96 Cm Curium 247	97 Bk Berkelium 247	98 Cf Californium 251	99 Es Einsteinium 254	100 Fm Fermium 257	101 Md Mendelevium 258	102 No Nobelium 255	103 Lr Lawrencium 256									